## **A New Phenoxyacetate-Based Linker System for the Solid-Phase Synthesis of Oligosaccharides**

## **ORGANIC LETTERS 2001 Vol. 3, No. 5 <sup>747</sup>**-**<sup>750</sup>**

**Xiangyang Wu, Matthias Grathwohl, and Richard R. Schmidt\***

*Fachbereich Chemie, Uni*V*ersita*¨*t Konstanz, Fach M 725, D-78457 Konstanz, Germany richard.schmidt@uni-konstanz.de*

**Received December 28, 2000**

## **ABSTRACT**



**A novel linker system has been designed, and its first application to solid-phase oligosaccharide synthesis is described. The use of the highly reactive** *o***-nitro-phenoxyacetate linker allows a fast and quantitative cleavage using mild basic conditions. This method combined with the** trichloroacetimidate glycosylation exhibits highly promising results as demonstrated for the synthesis of tetrasaccharide 1 ( $n = 3$ ) containing **glucose**  $\beta$ (1  $\rightarrow$  4) and  $\beta$ (1  $\rightarrow$  6) linkages.

The solid-phase synthesis of oligosaccharides has gained considerable interest during the last years<sup>1</sup> as a result of its possible advantages over conventional technologies, and many different linker systems have been developed.2 To employ the powerful trichloroacetimidate glycosylation method, $3$  temporary protective groups and the linker moiety have to be designed to be stable in the presence of catalytic

(3) (a) Schmidt, R. R. *Angew. Chem.* **<sup>1986</sup>**, *<sup>98</sup>*, 213-236; *Angew. Chem., Int. Ed. Engl.* **<sup>1986</sup>**, *<sup>25</sup>*, 212-235. (b) Schmidt, R. R.; Kinzy, W. *Ad*V*. Carbohydr. Chem. Biochem.* **<sup>1994</sup>**, *<sup>50</sup>*, 21-123.

10.1021/ol007062e CCC: \$20.00 © 2001 American Chemical Society **Published on Web 02/15/2001**

amounts of Lewis acid and to permit orthogonal cleavage under mild conditions. Most of the commonly used linker types in synthetic organic chemistry or peptide chemistry4 do not fulfill this demand. Therefore, the design of advanced protective groups and new efficient linker systems for the solid-phase synthesis of oligosaccharides remains a challenging task.

We present here our results concerning the preparation and first application of a new *o*-nitro-substituted phenoxyacetate linker system on Merrifield resin. Since it is known from solution-phase chemistry that an unsubstituted phenoxyacetate protecting group can be easily cleaved using methylamine, we chose the *o*-nitro-substituted derivative, which as a result of its electron-withdrawing effect should permit a particularly short reaction time for cleavage from the solid support.5 The system proved to meet all the requirements given above, leading to target tetrasaccharide molecule **1** in an excellent yield (Scheme 1). To the best of our knowledge the structural motif **1** was until now only

<sup>(1)</sup> For some recent reviews, see: (a) Haase, W. C.; Seeberger, P. H. *Curr. Org. Chem.* **<sup>2000</sup>**, *<sup>4</sup>*, 481-511. (b) Osborn, H. M. I.; Khan, T. H.

*Tetrahedron* **<sup>1999</sup>**, *<sup>55</sup>*, 1807-1850. (2) (a) Ester type linker: Roussel, F.; Knerr, L.; Grathwohl, M.; Schmidt, R. R. *Org. Lett.* **<sup>2000</sup>**, *<sup>2</sup>*, 3043-3046. Zhu, T.; Boons, G. J. *J. Am. Chem. Soc*. **<sup>2000</sup>**, *<sup>122</sup>*, 10222-10223. (b) Silyl ether type linker: Doi, T.; Sugiki, M.; Yamada, H.; Takahashi, T.; Porco, J. A., Jr. *Tetrahedron Lett.* **1999**, *<sup>40</sup>*, 2141-2144. Danishefsky, S. J.; McClure, K. F.; Randolph, J. T.; Ruggeri, R. B. *Science* **<sup>1993</sup>**, *<sup>260</sup>*, 1307-1309. (c) Thioether type linker: Rademann, J.; Geyer, A.; Schmidt, R. R. *Angew. Chem.* **<sup>1998</sup>**, *<sup>110</sup>*, 1309- 1313; *Angew. Chem., Int. Ed.* **<sup>1998</sup>**, *<sup>37</sup>*, 1241-1245. Yan, L.; Taylor, C. M.; Goodnow, R., Jr.; Kahne, D. *J. Am. Chem. Soc*. **<sup>1994</sup>**, *<sup>116</sup>*, 6953- 6954. (d) Benzyl ether type linker: Mehta, S.; Whitfield, D. M. *Tetrahedron Lett*. **<sup>1998</sup>**, *<sup>39</sup>*, 5907-5910. (e) Sulfone linker: Hunt, J. A.; Roush, W. R. *J. Am. Chem. Soc*. **<sup>1996</sup>**, *<sup>118</sup>*, 9998-9999. (f) Wang type linker: Manabe, S.; Ito, Y.; Ogawa, T. *Synlett* **<sup>1998</sup>**, 628-630. (g) Metathesis cleavable linker: Knerr, L.; Schmidt, R. R. *Eur. J. Org. Chem.* **<sup>2000</sup>**, 2803-2808. Andrade, R. B.; Plante, O. J.; Melean, L.; Seeberger, P. H. *Org. Lett.* **1999**, *<sup>1</sup>*, 1811-1814. (h) Selenium based linker: Nicolaou, K. C.; Mitchell, H. J.; Fylaktakidou, K. C.; Suzuki, H.; Rodrı´guez, R. M. *Angew. Chem.* **2000**, *<sup>112</sup>*, 1131-1135; *Angew. Chem., Int. Ed.* **<sup>2000</sup>**, *<sup>39</sup>*, 1089-1093.

<sup>(4)</sup> For a recent review on linkers, see: James, I. W. *Tetrahedron* **1999**, *<sup>55</sup>*, 4855-4946.

<sup>(5)</sup> For the use of this type of system in a reverse manner in oligosaccharide synthesis on Wang resin, see: Manabe, S.; Nakahara, Y.; Ito, Y. *Synlett* **<sup>2000</sup>**, *<sup>9</sup>*, 1241-1244.

<sup>(6)</sup> Kono, H.; Kawano, S.; Erata, T.; Takai, M. *J. Carbohydr. Chem.* **<sup>2000</sup>**, *<sup>19</sup>*, 127-140.



synthesized by an enzymatic approach under regioselective transglycosylation conditions in low yield.6

Retrosynthetic analysis of **1** led to the polymer-bound linker **3** and to the 9-fluorenylmethyloxycarbonyl (Fmoc) group bearing *O*-glucosyl trichloroacetimidate **2**. <sup>7</sup> Recently, we reported the efficient preparation of *O*-glycosyl trichloroacetimidates bearing Fmoc-protected hydroxy groups and their use as glycosyl donors both in solution and on solid phase.8 Other types of Fmoc-containing glycosyl donors have been recently also reported.9,10 Resin **3** led to the required building blocks **<sup>4</sup>**-**<sup>6</sup>** as well as to the activated trichloroacetimidate Merrifield resin **7**, <sup>11</sup> which was synthesized starting from commercially available hydroxymethyl-substituted Merrifield resin **8**. We describe here the first application of polymer reagent **7** in an acid-catalyzed benzyl ether synthesis. Compound **7** and all derivatives thereof exhibited higher stability against acids, as required for the solid-phase oligosaccharide synthesis, than the corresponding Wang resin.<sup>12</sup>

The synthesis of **1** begins with suitably protected known precursor **9**, <sup>13</sup> which is converted under reductive reaction

(7) Wu, X.; Schmidt, R. R. Unpublished results.

(11) Grathwohl, M. Dissertation, Universität Konstanz, Germany, 2001.

conditions into the corresponding primary alcohol **5** by a highly regioselective *p*-methoxy-benzylidene ring opening (Scheme 2).



 $a$  (a) 0.1 equiv Bu<sub>2</sub>BOTf, BH<sub>3</sub>·THF, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C. (b) DCC, DMAP,  $CH_2Cl_2$ , rt. (c) DDQ,  $CH_2Cl_2/H_2O$ , rt. (d) 2 equiv FmocCl, pyridine, rt. (e) HF'pyridine, THF, rt.

Starting from commercially available 4-hydroxy-3-nitrobenzaldehyde **10** the *o-*nitro-substituted phenoxyacetyl linker molecule **6** was prepared following a four-step literature procedure.5 Ester formation between primary alcohol **5** and carboxylic acid **6** was achieved by treatment with a catalytic amount of DMAP (0.1 equiv) and with DCC (2.0 equiv) as condensing agen, affording fully protected **11**. Cleavage of the MPM ether moiety of **11** was performed in the presence of DDQ (1.5 equiv) in a mixture of  $CH_2Cl_2/H_2O$  to obtain **12** in an excellent yield. Because of its high stability toward Lewis acid as well as its convenient cleavage under weak basic conditions, which can be readily followed by UV, for temporary protection the Fmoc group was employed. In our

<sup>(8)</sup> Roussel, F.; Knerr, L.; Grathwohl, M.; Schmidt, R. R. *Org. Lett.* **<sup>2000</sup>**, *<sup>2</sup>*, 3043-3046.

<sup>(9)</sup> For Fmoc-containing phosphoramidites, see: Freese, S. J.; Vann, W. F. Carbohydr. Res. 1996, 281, 313-319.

F. *Carbohydr. Res.* **<sup>1996</sup>**, *<sup>281</sup>*, 313-319. (10) For Fmoc-containing thioglycoside donors, see: (a) Nicolaou, K. C.; Winssinger, N.; Pastor, J.; DeRoose, F. *J. Am. Chem. Soc.* **1997**, *119*, <sup>449</sup>-450. (b) Nicolaou, K. C.; Watanabe, N.; Li, J.; Pastor, J.; Winssinger, N. *Angew. Chem.* **<sup>1998</sup>**, *<sup>110</sup>*, 1636-1638; *Angew. Chem., Int. Ed.* **<sup>1998</sup>**, *<sup>37</sup>*, 1559-1561. (c) Zhu, T.; Boons, G.-J. *Tetrahedron: Asymm.* **<sup>2000</sup>**, *<sup>11</sup>*, <sup>199</sup>-205.

<sup>(12)</sup> For Wang trichloroacetimidate resin, see: Hanessian, S.; Xie, F. *Tetrahedron Lett.* **<sup>1998</sup>**, *<sup>39</sup>*, 733-736.

<sup>(13)</sup> Joniak, K. *Chem. Z*V*esti* **<sup>1974</sup>**, 110-112.

<sup>(14)</sup> Trost, B. M.; Caldwell, C. G.; Murayama, E.; Heissler, D. *J. Org. Chem.* **<sup>1983</sup>**, *<sup>48</sup>*, 3252-3265.

<sup>(15)</sup> **Preparation of Trichloroacetimidate Resin 7.** After swelling of dry Merrifield resin 8 (10.0 g, 14.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (100 mL), trichloroacetonitrile (15 mL, 10 equiv) was added. The resulting suspension was cooled under argon to 0 °C and shaken for 10 min. DBU (0.44 mL, 0.2 equiv) was added, and the resulting mixture was shaken for 40 min under an inert gas atmosphere at 0 °C. The resin was rinsed off, switching three times between  $CH_2Cl_2$  and THF (each 100 mL), and dried under high vacuum to afford resin **7** (12.9 g, 14.6 mmol, quant).

hands this group has emerged as a very suitable protecting group for both solution- and solid-phase synthesis.<sup>8</sup> Installation of the Fmoc group to **12** using pyridine and FmocCl  $(\rightarrow 13)$  was followed by desilylation using 5 equiv of HF $\cdot$ py in THF14 to afford linker **4**, ready for loading onto the resin. Attachment of linker **4** to the hydroxymethylsubstituted Merrifield resin **8** was achieved via formation of trichloroacetimidate resin **7** and its activation with a catalytic amount of TMSOTf  $(0.3 \text{ equiv})$  at  $0 \text{ °C}$  (Scheme 3).



 $a$  (a) 10 equiv CCl<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C. (b) (i) 0.3 equiv TMSOTf,  $CH<sub>2</sub>Cl<sub>2</sub>$ , 0 °C. (ii) MeOH.

It is to be noted that conversion of **8** to **7** proceeds smoothly and very quickly in a quantitative manner. The reaction can be readily performed up to a 10-g scale, and the resulting activated trichloroacetimidate polymer reagent **7** showed a high stability and was storable for at least 6 months.15 After linker attachment affording resin **14** the remaining activated benzyl functions were quenched by direct addition of methanol. Loading of resin **14** was determined by a very fast and clean preparative cleavage of **15** by excess MeNH2 (10 min) followed by acetylation to be 0.213 mmol/g (Scheme 4).16 Under these conditions the linker was selectively cleaved without affecting any *O*-acetyl groups; loading was also determined by recycling unreacted linker molecule **4** from the washing solution.

Generation of polymer-bound acceptor **3** was performed by treatment with a mixture of dichloromethane and triethy-



 $a$  (a) (i) MeNH<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt. (ii) Ac<sub>2</sub>O/pyridine, rt. (b) NEt<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt. (c) 0.3 equiv TMSOTf, 4 Å, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C.

lamine until the generated UV spot completely disappeared.<sup>17</sup> Solid-phase glycosylation with the Fmoc-bearing *O*-glucosyl trichloroacetimidate 2 (3.0 equiv) at  $0^{\circ}$ C and repetition of the described sequence in a cyclic manner afforded resinbound oligosaccharides **16** ( $n = 1-3$ ) and **17** ( $n = 1, 2$ ).<sup>18</sup> All glycosylations were performed only once with the exception of the first one, which required repetition in order to reach a complete  $\beta(1 \rightarrow 4)$  linkage.<sup>19</sup>

The described solid-phase synthesis cycle was performed three times. After each run preparative cleavage of the polymer-bound structures **16** ( $n = 1-3$ ) using the conditions described above<sup>16</sup> furnished the corresponding oligosaccharides **18**, **19**, and **1** in excellent yields (Scheme 5). Cellobiose derivative **18**<sup>20</sup> was isolated in 85% overall yield over four steps starting from **14** (96% per step). Up to the tetrameric stage no significant drop in terms of efficiency of this new

<sup>(16)</sup> **General Procedure for Cleavage.** Dry resin was swollen in CH2-  $Cl<sub>2</sub>$  (10 mL/g resin), and the resulting suspension was shaken for 10 min under an inert gas atmosphere. A solution of ∼8 M MeNH2 in EtOH (∼100 equiv) was added, and the resulting mixture was shaken for 10 min under argon. The resin was rinsed off, switching three times between  $CH_2Cl_2$ and THF. The evaporated filtrates were treated by a mixture of acetic anhydride and pyridine  $(1:1, 20 \text{ mL/g})$  for 2 h. The resulting residues were purified by flash chromatography.

<sup>(17)</sup> **General Procedure for Deprotection.** Dry resin was swollen in a mixture of  $CH_2Cl_2/NEt_3$  (4:1), and the resulting suspension was shaken for 4 h. The resin was treated as described above and dried under high vacuum.

<sup>(18)</sup> **General Procedure for Glycosylation.** Dry acceptor loaded resin was directly swollen in a CH<sub>2</sub>Cl<sub>2</sub> solution (15 mL/g resin) containing donor **2** (3 equiv) and 4 Å molecular sieves. The resulting suspension was cooled under argon to 0 °C and shaken for 10 min. A solution of a freshly prepared  $0.5$  M TMSOTf solution in CH<sub>2</sub>Cl<sub>2</sub> (0.3 equiv) was added, and the resulting mixture was shaken for 1 h under an inert gas atmosphere. The resin was treated as described above.

<sup>(19)</sup> Solid-phase glycosylation to resin bound disaccharide **16** ( $n = 1$ ) was repeated once under the same conditions.

<sup>(20)</sup> **Analytical Data of Compound 18.** MALDI-TOF (DHB/THF): *m/z* calcd M (C<sub>32</sub>H<sub>42</sub>O<sub>17</sub>) 698.67; ( $\overrightarrow{M}$  + Na)<sup>+</sup> 721.66; ( $\overrightarrow{M}$  + K)<sup>+</sup> 737.77; found 721.9, 737.9. 1H NMR (600 MHz, CDCl3): *δ* 2.06 (m, 6 COC*H*3), 3.36 (s, OC*H*<sub>3</sub>), 3.55 (m, 5b-*H*), 3.65 (t,  $3J = 3.3$  Hz, 4b-*H*), 3.71 (t,  $3J = 3.2$  Hz,  $4a$ -*H*), 3.87 (m, 5a-*H*), 4.13 (dd,  $J_{6.6'} = 12.0$  Hz,  $J_{5.6} = 4.8$  Hz, 6a-*H*), 4.27  $(2d, {}^{3}J = 3.7 \text{ Hz}, 6'b - H, 6b - H)$ , 4.49  $(d, J_{1,2} = 7.9 \text{ Hz}, 1b - H)$ , 4.50  $(d, {}^{3}J)$  $=$  3.0 Hz, 6<sup> $\alpha$ </sup>a-*H*), 4.53–4.59 (m, C*H*<sub>2</sub>Ph), 4.80 (dd,  $J_{1,2} = 3.7$  Hz,  $J_{2,3} =$ 10.2 Hz, 2a-*H*), 4.84 (m, 2b-*H*), 4.85 (d, *<sup>J</sup>*1,2 ) 3.7 Hz, 1a-*H*), 5.19 (t, <sup>3</sup>*<sup>J</sup>*  $= 9.2$  Hz, 3b-*H*), 5.43 (t,  $3J = 9.7$  Hz, 3a-*H*), 7.21-7.31 (m, *Ar*). <sup>13</sup>C NMR (150.9 MHz, CDCl<sub>3</sub>):  $\delta = 62.2$  (C-6a), 63.1 (C-6b), 65.4 (C-5a), 70.1 (C-(150.9 MHz, CDCl3): *<sup>δ</sup>* ) 62.2 (*C*-6a), 63.1 (*C*-6b), 65.4 (*C*-5a), 70.1 (*C*-3a), 71.3 (*C*-2a), 72.5 (*C*-2b), 73.3 (*C*-5b), 75.5 (*C*-4b), 75.7 (*C*-3b), 77.0 (*C*-4a), 97.1 (*C*-1a), 101.1 (*C*-1b).

<sup>(21)</sup> **Analytical Data of Compound 19.** MALDI-TOF (DHB/THF): *m/z* calcd M (C<sub>49</sub>H<sub>62</sub>O<sub>24</sub>) 1035.00;  $(M + Na)^+$  1057.99;  $(M + K)^+$  1074.10; found 1057.3, 1073.3. 1H NMR (600 MHz, CDCl3): *δ* 2.03 (m, 8 COC*H*3), 3.36 (s, OC*H*<sub>3</sub>), 3.40 (m, 5b-*H*), 3.55 (m, 5c-*H*), 3.62 (m, 4b-*H*, 4c-*H*), 3.66 (m, 6b-*H*), 3.69 (m, 4a-*H*), 3.76 (d, <sup>3</sup>*J* = 7.3 Hz, 5a-*H*), 4.10 (m, 6'b-*H*) 4.13 (d, <sup>3</sup>*J* = 5.0 Hz, 6a-*H*), 4.17 (dd, *J<sub>s* </sub> 6′b-*H*), 4.13 (d, <sup>3</sup>*J* = 5.0 Hz, 6a-*H*), 4.17 (dd,  $J_{6,6'} = 12.0$  Hz,  $J_{5,6} = 4.6$  Hz, 6c-*H*), 4.34 (d, <sup>2</sup>*J* = 2.2 Hz, 6′c-*H*), 4.44 (d, *J*<sub>1.2</sub> = 7.9 Hz, 1b-*H*), 4.48 Hz, 6c-*H*), 4.34 (d, <sup>2</sup>*J* = 2.2 Hz, 6<sup>'</sup>c-*H*), 4.44 (d, *J*<sub>1,2</sub> = 7.9 Hz, 1b-*H*), 4.48<br>(d, *J*<sub>1,2</sub> = 7.9 Hz, 1c-*H*), 4.49 (d, <sup>2</sup>*J* = 2.1 Hz, 6'a-*H*), 4.53–4.59 (m, 2) (d,  $J_{1,2} = 7.9$  Hz, 1c-*H*), 4.49 (d,  $^2J = 2.1$  Hz, 6′a-*H*), 4.53-4.59 (m, 2<br>C*H*<sub>2</sub>Ph) 4.80 (dd,  $J_{1,2} = 7.9$  Hz,  $J_{2,3} = 6.2$  Hz, 2b-*H*), 4.81 (m, 2a-*H*)  $CH_2Ph$ , 4.80 (dd,  $J_{1,2} = 7.9$  Hz,  $J_{2,3} = 6.2$  Hz, 2b-*H*), 4.81 (m, 2a-*H*),



 $a$  (a) (i) MeNH<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt. (ii) Ac<sub>2</sub>O/pyridine, rt.

method was observed. Thus, trisaccharide **19**<sup>21</sup> (68% overall over six steps, average 95% per step) and tetrasaccharide **1**<sup>22</sup> (34% overall over eight steps, average 93% per step) were isolated in similar remarkable yields.

In summary, we have presented the synthesis of a new highly reactive linker system and its first successful application to solid-phase oligosaccharide synthesis. This novel *o*-nitro-phenoxyacetate linker proved to be completely stable under glycosylation and Fmoc removal conditions and could be cleaved very quickly at the end of the synthesis using only volatile reagents. Thus, Fmoc as the temporary hydroxy protecting group of *O*-glycosyl trichloroacetimidates was found to be a very powerful combination leading to tetrasaccharide **1** in an excellent overall yield. Currently we are developing the application of this new linker system toward higher oligomers and more complex carbohydrate structures.

**Acknowledgment.** We thank the European Community  $(grant no. FAIR-CT97-3142)$ , the Bundesministerium für Bildung und Forschung (grant no. 0311 229) and the Deutsche Forschungsgemeinschaft for financial support of this work. We are grateful to Dr. A. Geyer for his help in the structural assignments.

OL007062E

<sup>4.86 (</sup>d,  $J_{1,2} = 3.5$  Hz, 1a-*H*), 4.89 (m, 2c-*H*), 5.17 (t, <sup>3</sup>J = 9.3 Hz, 3b-*H*), 5.23 (t, <sup>3</sup>J = 9.4 Hz, 3c-*H*), 5.42 (t, <sup>3</sup>J = 9.7 Hz, 3a-*H*), 7.22–7.32 (m, 5.23 (t, <sup>3</sup>*<sup>J</sup>* ) 9.4 Hz, 3c-*H*), 5.42 (t, <sup>3</sup>*<sup>J</sup>* ) 9.7 Hz, 3a-*H*), 7.22-7.32 (m, *Ar*). 13C NMR (150.9 MHz, CDCl3): *δ* 62.2 (*C*-6a), 62.8 (*C*-6c), 68.1 (*C*-6b), 70.0 (*C*-3a), 71.4 (*C*-2a), 72.3 (*C*-2c), 72.7 (*C*-2b), 73.4 (*C*-5c), 74.7 (*C*-5b), 75.6 (*C*-3c), 75.7 (*C*-3b), 76.0 (*C*-4b, *C*-4c), 77.0 (*C*-4a), 97.0 (*C*-1a), 100.9 (*C*-1c), 101.3 (*C*-1b).

<sup>(22)</sup> **Analytical Data of Compound 1.** MALDI-TOF (DHB/THF): *m/z* calcd M (C<sub>66</sub>H<sub>82</sub>O<sub>31</sub>) 1371.34; (M + Na)<sup>+</sup> 1394.33; (M + K)<sup>+</sup> 1410.44; found 1393 7 1410.3<sup>-1</sup>H NMR (600 MHz CDCL);  $\delta$  2.02 (m 11 COCH) found 1393.7, 1410.3. 1H NMR (600 MHz, CDCl3): *δ* 2.02 (m, 11 COC*H*3), 3.35 (s, OC*H*3), 3.48 (m, 4c-*H*), 3.50 (m, 5b-*H*), 3.51 (m, 5c-*H*), 3.58 (m, 5d-*H*), 3.64 (d, *J* = 2.3 Hz, 6c-*H*), 3.65 (d, *J* = 2.4 Hz, 6b-*H*), 3.67 (m, 4b-*H*), 3.68 (m, 4d-*H*), 3.72 (d, <sup>3</sup>*J* = 9.6 Hz, 4a-*H*), 3.85 (m, 5a-*H*), 4.03 (d. <sup>2</sup>*J* = 10.8 Hz, 6<sup>4</sup>)-*H*), 4.12 (t. 6a-*H*), 4.2  $(d, {}^{2}J = 10.8 \text{ Hz}, 6'b - H)$ , 4.12 (t, 6a-*H*), 4.20 (m, 6'c-*H*, 6d-*H*), 4.37 (dd,  $J_{6,6'} = 12.0$  Hz,  $J_{5,6'} = 2.1$  Hz, 6′d-*H*), 4.46 (m, 1c–H, 6′a-*H*), 4.49 (d,  $J_{1,2} = 8.0$  Hz, 1b-*H*), 4.52–4.59 (m, 3 C*H*<sub>2</sub>Ph), 4.58 (m, 1d-*H*), 4.81 (d,  $J_{1,2} =$ ) 8.0 Hz, 1b-*H*), 4.52-4.59 (m, 3 C*H2*Ph), 4.58 (m, 1d-*H*), 4.81 (d, *<sup>J</sup>*1,2 ) 3.2 Hz, 1a-*H*), 4.82 (m, 2b-*H*), 4.84 (m, 2c-*H*), 4.86 (m, 2a-*H*), 4.89 (m, 2d-*H*), 5.18 (m, 3b-*H*), 5.20 (m, 3c-*H*), 5.21 (m, 3d-*H*), 5.40 (t, <sup>3</sup>*J* = 9.7<br>Hz, 3a-*H*), 7.22–7.35 (m, *Ar*), <sup>13</sup>C NMR (150.9 MHz, CDCl<sub>3</sub>); δ.62.3 Hz, 3a-*H*), 7.22-7.35 (m, *Ar*). 13C NMR (150.9 MHz, CDCl3): *<sup>δ</sup>* 62.3 (*C*-6a), 62.9 (*C*-6d), 67.6 (*C*-6c), 68.5 (*C*-6b, *C*-5a), 70.1 (*C*-3a), 71.2 (*C*-2a), 72.1 (*C*-2d), 72.4 (*C*-2c), 72.9 (*C*-2b), 73.6 (*C*-5d), 74.2 (*C*-5b), 75.3 (*C*-5c, *C*-3d), 75.6 (*C*-3c), 75.8 (*C*-3, *C*-4d), 75.9 (*C*-4b), 76.7 (*C*-4a, *C*-4c), 96.9 (*C*-1a), 100.8 (*C*-1c), 101.1 (*C*-1b), 101.4 (*C*-1d).